

1,3-Carbothiolation of 4-(Trifluoromethyl)-*p*-Quinols: A New Access to Functionalized (Trifluoromethyl)arenes

Xiao Liu, Ling Pan,* Jinhuan Dong, Xianxiu Xu, Qian Zhang, and Qun Liu*

Department of Chemistry, Northeast Normal University, Changchun 130024, China

panl948@nenu.edu.cn; liuqun@nenu.edu.cn

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ABSTRACT



A new strategy, the *1,3-carbothiolation/aromatization*, for the synthesis of functionalized (trifluoromethyl)arenes has been developed that enables the regioselective introduction of two different functional groups onto an “aromatic ring” in the *meta*-position to each other in a single step.

Organofluorine compounds have received increasing interest because of their higher bioavailability compared to that of the corresponding nonfluorinated analogues.^{1,2} Thus, over the past two decades, (trifluoromethyl)trimethylsilane (TMSCF₃)³ has been widely applied since its first synthetic application as a nucleophilic trifluoromethylating reagent.^{3b}

Different from numerous methods for the synthesis of trifluoromethylated arenes that rely on substitution of a preexisting aromatic ring,⁴ in 1989, Stahly and Bell described the monotrifluoromethylation of *p*-quinones to give 4-(trifluoromethyl)-*p*-quinols^{5a} as precursors of (trifluoromethyl)arenes.⁵ More recently, Hu and co-workers revealed an

unprecedented *vicinal* trifluoromethylation/iodination of arynes, which can introduce CF₃ and I groups onto an aromatic ring in a single step.⁶ Herein, a new reaction, the *1,3-carbothiolation/aromatization* of 4-CF₃-*p*-quinol derivatives **1** (Figure 1), is described (Scheme 1). This reaction allows the regioselective introduction of two different functional groups onto an aromatic ring in the *meta*-position to each other (*meta*-double functionalization) and therefore enables the construction of functionalized CF₃-arenes^{4–6} in a single operation with readily available **1** as the nonaromatic precursors and ketene dithioacetals **2** (Figure 1). The latter are versatile intermediates in organic synthesis,⁷ both as carbon and sulfur nucleophiles.^{7a,8a,8b}

In the present research, the model reaction of 4-(trifluoromethyl)-*p*-quinol silyl ether **1a** with 3,3-bis(methylthio)-1-phenylprop-2-en-1-one **2a** (Figure 1) was first investigated using CuBr₂, Cu(OAc)₂, SnCl₄·5H₂O, SnCl₄, AlCl₃, TfOH (triflic acid), Pd(OAc)₂, BF₃·Et₂O,

(1) (a) Purser, S.; Moore, P. R.; Swallow, S.; Gouverneur, V. *Chem. Soc. Rev.* **2008**, 37, 320. (b) Müller, K.; Faeh, C.; Diederich, F. *Science* **2007**, 317, 1881. (c) Furuya, T.; Kamlet, A. S.; Ritter, T. *Nature* **2011**, 473, 470.

(2) (a) Prakash, G. K. S.; Mandal, M. *J. Fluorine Chem.* **2001**, 112, 123. (b) Singh, R. P.; Shreeve, J. M. *Tetrahedron* **2000**, 56, 7613. (c) Ma, J.-A.; Cahard, D. *J. Fluorine Chem.* **2007**, 128, 975. (d) Prakash, G. K. S.; Jog, P. V.; Batamack, P. T. D.; Olah, G. A. *Science* **2012**, 338, 1324.

(3) (a) Ruppert, I.; Schlich, K.; Volbach, W. *Tetrahedron Lett.* **1984**, 25, 2195. (b) Prakash, G. K. S.; Krishnamuti, R.; Olah, G. A. *J. Am. Chem. Soc.* **1989**, 111, 393.

(4) (a) Tomashenko, O. A.; Grushin, V. V. *Chem. Rev.* **2011**, 111, 4475. (b) Nagib, D. A.; MacMillan, D. W. C. *Nature* **2011**, 480, 224.

(5) (a) Stahly, P. G.; Bell, R. D. *J. Org. Chem.* **1989**, 54, 2873. (b) Stahly, G. P.; Jackson, A. *J. Org. Chem.* **1991**, 56, 5472. (c) Singh, R.; Czekelius, C.; Schrock, R. R.; Muller, P.; Hoveyda, A. H. *Organometallics* **2007**, 26, 2528. (d) Large, S.; Roques, N.; Langlois, B. R. *J. Org. Chem.* **2000**, 65, 8848. (e) Radix-Large, S.; Kucharski, S.; Langlois, B. R. *Synthesis* **2004**, 456.

(6) Zeng, Y.; Zhang, L.; Zhao, Y.; Ni, C.; Zhao, J.; Hu, J. *J. Am. Chem. Soc.* **2013**, 135, 2955 (10 examples, 35–86% yields), using excess [AgCF₃], CsF, TMP (2,2,6,6-tetramethylpiperidine), and 1-iodophenylacetylene (as iodine source).

(7) For recent reviews, see: (a) Pan, L.; Bi, X.; Liu, Q. *Chem. Soc. Rev.* **2013**, 42, 1251. (b) Pan, L.; Liu, Q. *Synlett* **2011**, 1073. For recent reports, see: (c) Xu, X.; Zhang, L.; Liu, X.; Pan, L.; Liu, Q. *Angew. Chem., Int. Ed.* **2013**, 52, 9271. (d) Li, Y.; Xu, X.; Tan, J.; Xia, C.; Zhang, D.; Liu, Q. *J. Am. Chem. Soc.* **2011**, 133, 1775. (e) Lee, J.-W.; List, B. *J. Am. Chem. Soc.* **2012**, 134, 18245.

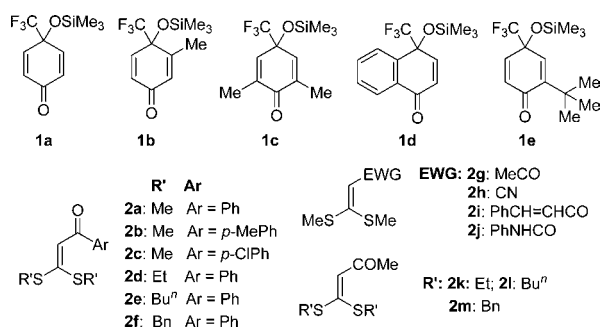


Figure 1. Selected 4-(trifluoromethyl)-*p*-quinols and ketene dithioacetals.

and In(OTf)₃ as the catalysts, respectively.^{7,8} After careful screening of the reaction conditions (for details, see the Supporting Information), it was found that, when catalyzed by In(OTf)₃, a mixture of (trifluoromethyl)arenes **3aa/3'aa** (in the tautomeric keto–enol form) were obtained in 95% overall yields by reaction in DCE (1,2-dichloroethane, Scheme 1), whereas, in the presence of TMSCl (trimethylsilyl chloride) and catalyzed by In(OTf)₃, (trifluoromethyl)arene **4aa** was produced in 92% yield (Scheme 1). However, under identical conditions for 6 h but promoted only by TMSCl (2.0 equiv), no reaction occurred for reaction of **1a** with **2a** (Table S1, entry 14, Supporting Information), and no transformation of **3aa/3'aa** to **4aa** was observed. These results indicate that TMSCl can accelerate the reaction.

The successful construction of CF₃-bearing aryl sulfides **3aa/3'aa** and **4aa** significantly extends the synthetic utility of the readily available 4-(trifluoromethyl)-*p*-quinol derivatives⁵ and adds a new entry to the chemo- and regioselective construction of polyfunctionalized arenes from readily available nonaromatic precursors.^{4–6,9} Therefore, the scope of the *meta*-double-functionalization of 4-CF₃-*p*-quinol derivatives **1** with ketene dithioacetals **2** was next examined. As a result, under optimal conditions (Scheme 1, for products **3aa/3'aa** and **4aa**), a series of functionalized CF₃-arenes was prepared, and the results are summarized in Scheme 1.

According to the experimental results (Scheme 1), methyl aryl sulfides **3aa/3'aa–3ac/3'ac** with phenacyl substituents on the aryl ring were obtained in excellent yields from reactions of **1a** with **2a–c**, respectively. Similarly, alkyl aryl sulfides **3ad/3'ad** and **3af/3'af** with ethylthio and benzylthio groups were prepared in high yields from reactions of **1a** with **2d** and **2f**. Similarly, sulfides **3ag** and **3'ak–3'am** were prepared in high yield from the reactions of **1a** with the corresponding α -acetyl ketene dithioacetals **2g** and **2k–m** having two methyl, ethyl,

benzyl, or *n*-butyl groups on the ketene dithioacetyl moiety. It was noted that the reaction of **1a** with α -cinnamoyl ketene dithioacetal **2i** showed a good tolerance for the cinnamoyl group giving **3'ai** in high yield. Moreover, all reactions were completed within 0.4–2.4 h except for the reaction of **1a** with α -cyano ketene dithioacetal **2h**, which led to *p*-quinol and 4-hydroxy-4-(trifluoromethyl)cyclohexa-2,5-dienone in 70% yield, instead of **3ah**. Thus, the desired sulfide **3ah** was prepared under identical conditions as above but catalyzed by SnCl₄·5H₂O. In products **3/3'**, **3ag**, **3'ai**, and **3'ak–3'am** exist primarily in their enol forms, whereas others, except for **3ah**, in the tautomeric keto–enol forms.¹⁰

All of the functionalized (trifluoromethyl)arenes **3/3'** show that an (alkylthiocarbonyl)methyl and an alkylthio group can be introduced into the *ortho* and *para* positions of the electron-deficient (trifluoromethyl)benzene ring. This efficient domino sequence comprises a fundamentally new way to synthesize (trifluoromethyl)arenes having high structural complexity.^{1,4–6} Fortunately, the scope of the synthesis of functionalized (trifluoromethyl)arenes **4** under optimized conditions (Scheme 1, for products **4aa**) gave further powerful support for the efficiency of the double functionalization/aromatization reaction.

The corresponding functionalized (trifluoromethyl)arenes **4aa–ag**, **4ai**, **4aj**, **4ba**, **4ca**, **4da**, **4dg**, **4ea**, and **4ak–am** were synthesized in high-to-excellent yields (except for **4ea** in 40% yield due to the steric hindrance of the bulky *t*-Bu group) under similar reaction conditions as the synthesis of **3** but in the presence of TMSCl (Scheme 1). Similar to the preparation of **3ah**, **4ah** was obtained in high yield under identical conditions but catalyzed by SnCl₄·5H₂O in the absence of TMSCl. For the formation of CF₃-arenes **4**, the electron-donating alkyl substituents on the *p*-quinol ring of substrates **1b**, **1c**, and **1e** did not affect the regioselectivity of the reaction (Scheme 1). In addition, 4-(trifluoromethyl)-4-(trimethylsilyloxy)naphthalen-1(4*H*)-one **1d** was also proved to be the suitable substrate (Scheme 1, products **4da** and **4dg**). Clearly, our new method based on the tandem 1,3-carbothiolation/aromatization sequence is flexible and opens up new possibilities for the construction of valuable polyfunctionalized arenes.

Thus, (trifluoromethyl)arenes **3** and **4**, which also share structural features of aryl sulfides, α -aryl ketones, α -aryl nitriles, and related structural motif,^{11–14} are obtained selectively under controllable conditions with or without the addition of TMSCl. In addition, products **3** also have the β -ketothioester motif, allowing straightforward access, in principle, to a series of functional groups such as ketones,

(10) CCDC 906801 (**3'aa**) and CCDC 896680 (**4aa**) contain the supplementary crystallographic data for this paper. These data can be obtained free of charge from the Cambridge Crystallographic Data Centre via www.dcd.ccr.cam.ac.uk/data_request/cif.

(11) (a) Beletskaya, I. P.; Ananikov, V. P. *Chem. Rev.* **2011**, *111*, 1596. (b) Jiang, C.-S.; Muller, W. E. G.; Schroder, H. C.; Guo, Y.-W. *Chem. Rev.* **2012**, *112*, 2179. (c) Hao, G.-F.; Wang, F.; Li, H.; Zhu, X.-L.; Yang, W.-C.; Huang, L.-S.; Wu, J.-W.; Berry, E. A.; Yang, G.-F. *J. Am. Chem. Soc.* **2012**, *134*, 11168.

(12) Culkin, D. A.; Hartwig, J. F. *Acc. Chem. Res.* **2003**, *36*, 234.

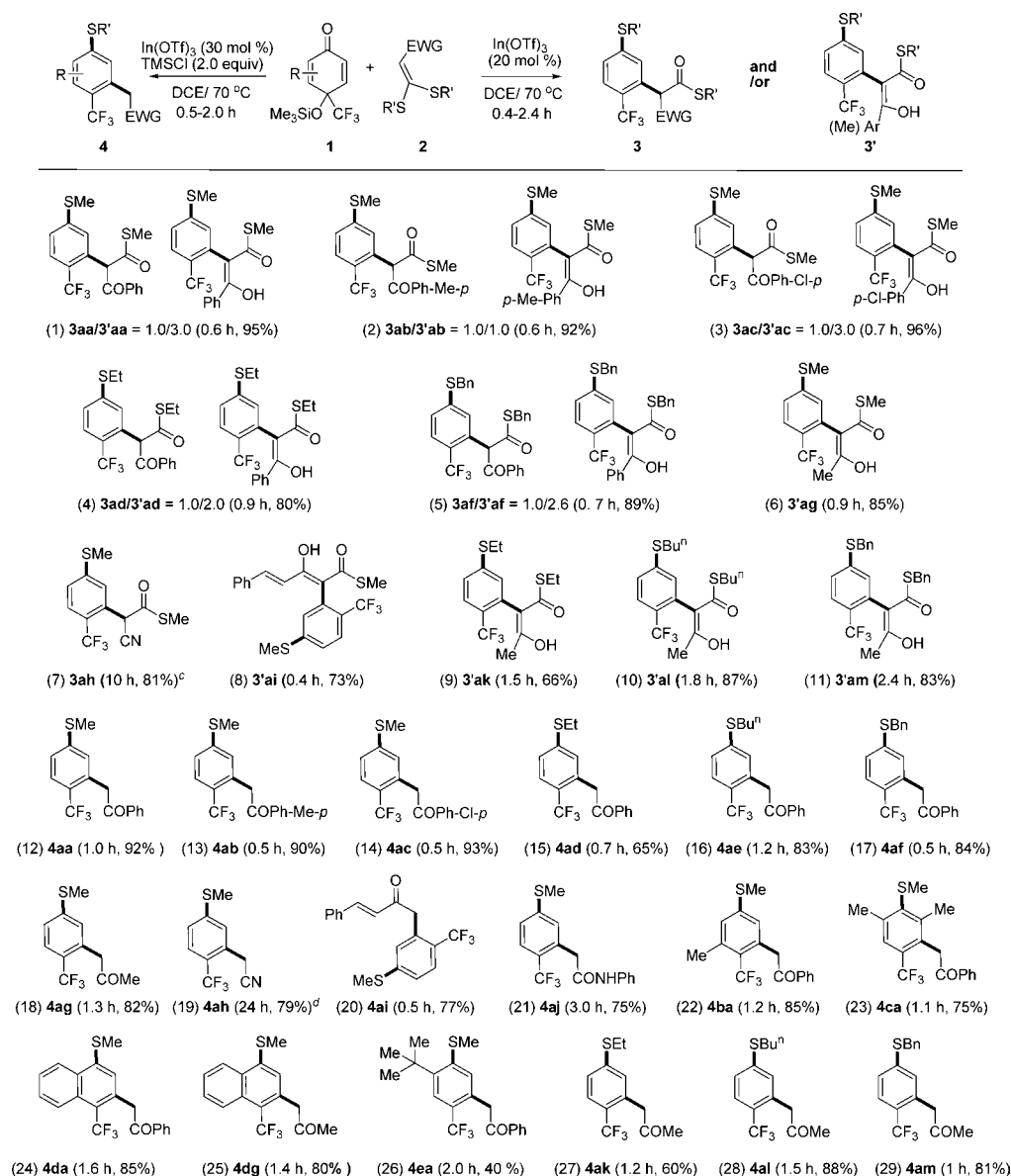
(13) Bellina, F.; Rossi, R. *Chem. Rev.* **2010**, *110*, 1082.

(14) Johansson, C. C. C.; Colacot, T. J. *Angew. Chem., Int. Ed.* **2010**, *49*, 676.

(8) (a) Liu, Y.; Liu, J.; Wang, M.; Liu, J.; Liu, Q. *Adv. Synth. Catal.* **2012**, *354*, 2678. (b) Xu, C.; Liu, J.; Ming, W.; Liu, Y.; Liu, J.; Wang, M.; Liu, Q. *Chem.—Eur. J.* **2013**, *19*, 9104. (c) Liu, X.; Xu, X.; Pan, L.; Zhang, Q.; Liu, Q. *Org. Biomol. Chem.* **2013**, *11*, 6703.

(9) Izawa, Y.; Pun, D.; Stahl, S. S. *Science* **2011**, *333*, 209.

Scheme 1. Synthesis of Functionalized (Trifluoromethyl)arenes^{a,b}



^a Conditions for **3/3'**: **1** (0.60 mmol), **2** (0.50 mmol), $\text{In}(\text{OTf})_3$ (20 mol %), DCE (2 mL), 70 °C, 0.4–2.4 h. ^b Conditions for **4**: **1** (0.60 mmol), **2** (0.50 mmol), $\text{In}(\text{OTf})_3$ (30 mol %), TMSCl (2.0 equiv), DCE (2 mL), 70 °C, 0.5–2.0 h. ^c Catalyzed by $\text{SnCl}_4 \cdot 5\text{H}_2\text{O}$ (30 mol %) and reacted for 10 h. ^d Catalyzed by $\text{SnCl}_4 \cdot 5\text{H}_2\text{O}$ (30 mol %) in THF (2.0 mL) in the absence of TMSCl and reacted for 24 h.

aldehydes, esters, amides and α -hydroxy- β -ketothioesters upon a single transformation.^{15,16} Fortunately, the chemo-selective oxidation of **4ag**, **4dg**, and **4am** to the corresponding aryl aldehydes **5** was also successful (Scheme 2).¹⁷

On the basis of the above results (Scheme 1), a mechanism for the formation of **3** and **4** is proposed (Scheme 3, with **3aa** and **4aa** as an example). The overall process involves (1) hydrolysis of ketene dithioacetal **2a** to release

methanethiol under acidic conditions in the presence of water (moisture);^{7a,e,18} (2) formation of complex **I** from **1a**, $\text{In}(\text{OTf})_3$ and methanethiol;^{8a} (3) addition of methanethiol

(15) Baidya, M.; Griffin, K. A.; Yamamoto, H. *J. Am. Chem. Soc.* **2012**, *134*, 18566.

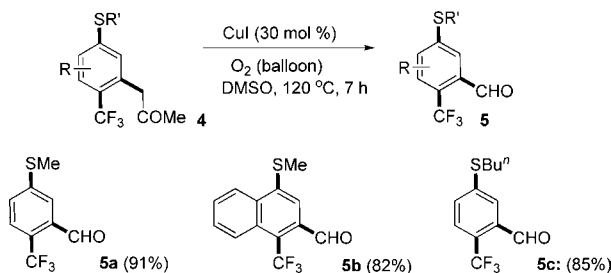
(16) Hatano, M.; Moriyama, K.; Maki, T.; shihara, K. *Angew. Chem., Int. Ed.* **2010**, *49*, 3823.

(17) Zhang, L.; Bi, X.; Guan, X.; Li, X.; Liu, Q.; Barry, B.-D.; Liao, P. *Angew. Chem., Int. Ed.* **2013**, *52*, 11303.

(18) (a) Liu, Q.; Che, G.; Yu, H.; Liu, Y.; Zhang, J.; Zhang, Q.; Dong, D. *J. Org. Chem.* **2003**, *68*, 9148. (b) Dong, D.; Ouyang, Y.; Yu, H.; Liu, Q.; Liu, J.; Wang, M.; Zhu, J. *J. Org. Chem.* **2005**, *70*, 4535. (c) Joshi, G.; Anslyn, E. V. *Org. Lett.* **2012**, *14*, 4714. (d) Truong, V. X.; Dove, A. P. *Angew. Chem., Int. Ed.* **2013**, *52*, 4132.

(19) (a) Hansen, A. M.; Lindsay, K. B.; Antharjanam, P. K. S.; Karaffa, J.; Daasbjerg, K.; Flowers, R. A.; II; Skrydstrup, T. *J. Am. Chem. Soc.* **2006**, *128*, 9616. (b) Wada, Y.; Otani, K.; Endo, N.; Kita, Y.; Fujioka, H. *Chem. Commun.* **2010**, *46*, 797.

(20) (a) Sloman, D. L.; Mitasev, B.; Scully, S. S.; Beutler, J. A.; Porco, J. A., Jr. *Angew. Chem., Int. Ed.* **2011**, *50*, 2511. (b) Dohi, T.; Washimi, N.; Kamitanaka, T.; Fukushima, K.; Kita, Y. *Angew. Chem., Int. Ed.* **2011**, *50*, 6142.

Scheme 2. Synthesis of (Trifluoromethyl)aryl Aldehydes **5**

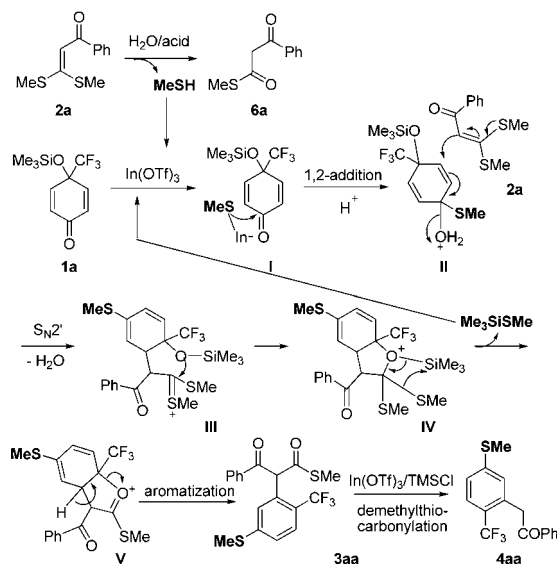
at the carbonyl group of **I** in a pseudointramolecular manner to give intermediate **II**;^{8a,14,19} (4) attack of the nucleophilic α -C of **2a** at **II** in a S_N2' manner,^{7a,b,8a,20} along with the release of water to produce thionium **III**; and finally, (5) further transformation involving intramolecular thiophilic attack of the silyloxy oxygen (**III** to **IV**),²¹ elimination of trimethyl(methylthio)silane (**IV** to **V**) and subsequent aromatization to give **3aa** or further to give **4aa** in the assistance of TMSCl.

Clearly, the above transformation would benefit from reincorporation of the ketene dithioacetals **2** into the products in the form of an efficient recycling of the alkylthiol and trimethyl(methylthio)silane generated in situ to enable further transformations. Accordingly, the synthesis of functionalized CF₃-arenes described in this study represents a new carbothiolation reaction,^{22,23} the catalytic intermolecular 1,3-carbothiolation, through *meta*-double functionalization of the nonaromatic precursors. Significantly, the 1,3-carbothiolation enables two different functional groups to be introduced into the aromatic ring in the *ortho* and *para* positions to an electron-withdrawing trifluoromethyl group, which has been a tough challenge in chemistry.

(21) Hoye, T. R.; Baire, B.; Niu, D.; Willoughby, P. H.; Woods, B. P. *Nature* **2012**, 490, 208.

(22) (a) Hooper, J. F.; Chaplin, A. B.; González-Rodríguez, C.; Thompson, A. L.; Weller, A. S.; Willis, M. C. *J. Am. Chem. Soc.* **2012**, 134, 2906. (b) Sugoh, K.; Kuniyasu, H.; Sugae, T.; Ohtaka, A.; Takai, Y.; Tanaka, A.; Machino, C.; Kambe, N.; Kurosawa, H. *J. Am. Chem. Soc.* **2001**, 123, 5108. (c) Nakamura, I.; Sato, T.; Yamamoto, Y. *Angew. Chem., Int. Ed.* **2006**, 45, 4473.

(23) (a) Denmark, S. E.; Jaunet, A. *J. Am. Chem. Soc.* **2013**, 136, 6419. (b) Fang, Z.; Yuan, H.; Liu, Y.; Tong, Z.; Li, H.; Yang, J.; Barry, B.-D.; Liu, J.; Liao, P.; Zhang, J.; Liu, Q.; Bi, X. *Chem. Commun.* **2012**, 48, 8802.

Scheme 3. Proposed Mechanism for Formation of **3** and **4**

In summary, a new strategy for the synthesis of functionalized trifluoromethyl arenes with readily available 4-(trifluoromethyl)-*p*-quinol derivatives as the nonaromatic precursors has been developed. The reactions can be carried out under mild reaction conditions to afford a series of (trifluoromethyl)arenes with important structural features, including aryl sulfides, α -arylated carbonyl and nitrile compounds, and β -ketothioesters in good to excellent yields in most cases in a single operation via a tandem 1,3-carbothiolation/aromatization sequence. Further investigations to demonstrate the utility of the intermolecular 1,3-carbothiolation reaction are in progress.

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Supporting Information Available. Experimental procedures and analytical and spectral data for all the new compounds. CCDC 906801 (**3'aa**) and CCDC 896680 (**4aa**). This material is available free of charge via the Internet at <http://pubs.acs.org>.

The authors declare no competing financial interest.